

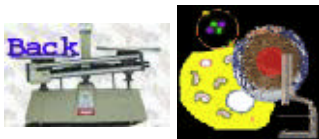
"...And so...all elements are made of atoms.
•Atoms are invisible, indivisible particles.
•Atoms of the same element are identical.
•Atoms of different elements are dissimilar.
•When two or more elements chemically combine they form a compound.

These are the critical parts of MY "atomic theory"!

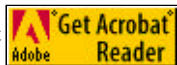


John Dalton (1803), an English meteorologist, used his knowledge of gases in the atmosphere as a model for the atom. He is given credit for proposing the first "Atomic Theory":

- All elements are composed of atoms.
- Atoms are indivisible, invisible particles.
- Atoms of the same element are the same.
- Atoms of different elements are different.
- Compounds are formed by two or more atoms joined together.



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Updated June 14, 2000 by: [Glen Westbrook](#)

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